





# Which is the Best Fuel?

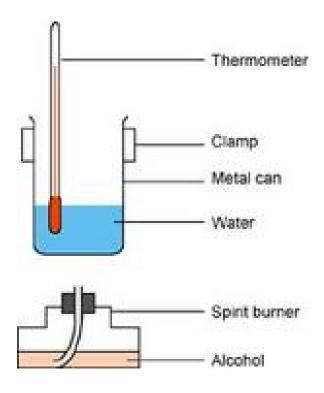
# **Teacher Notes**

## Possible meaning of 'best fuel'

- 1. The best fuel is the one that heats the water the fastest.
- 2. The best fuel is the one of which the least amount is used to heat the water
- 3. The best fuel is the one that is most economical, in terms of cost, to heat the water.

# Student Worksheet 1

## Experimental Guidelines to Determine the 'Best' Fuel



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#### Procedure

- 1. Half fill the spirit burner with the fuel supplied. (This may be done for you already).
- 2. Weigh the spirit burner containing the fuel.
- 3. Set up the apparatus for burning the fuel and heating the water as in the diagram.
- 4. Measure out  $200 \text{ cm}^3$  water and pour into the can to be heated.
- 5. Take the temperature of the water in the can.
- 6. Light the burner and at the same time start the stop clock.
- 7. Put draught shields around the can and burner (if necessary).
- 8. Stir the water slowly.
- 9. Note the temperature of the water by reading the thermometer at half minute intervals.
- 10. When the water is at about 50°C, extinguish the burner.
- 11. Stop the stop-clock and record the time for which the fuel was burning
- 12. Continue to stir the water in the can and then record the highest water temperature reached. Determine the temperature rise of the water.
- 13. Weigh the burner + the remaining fuel.
- 14. Calculate the amount of fuel used in the experiment.

Repeat the experiment with other fuels if instructed to do so by the teacher.

# Student Handout 2

### Recording of Results and Giving an Interpretation

#### Results

Complete the following table

Fuel 1 Fuel 2 Fuel 3

Amount of fuel used for a 10°C rise in water temperature

Time taken for a 10°C rise in water temperature

Cost of the fuel used

#### Interpretation

The amount of fuel used for the  $10^{\circ}$ C rise in temperature is a measure of the heat of combustion of the fuel. (In textbooks this value is standardised using internationally agreed specifications –  $1^{\circ}$ C rise per mole)

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Decide on the 'best' fuel.		
	I suggest the best fuel is	
	My reason for this is	

I think the calorific value or heat of combustion of a fuel (is/is not) a good measure of the 'best' fuel.

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